Code: BP701T

B.Pharm IV Year I Semester (R19) Regular Examinations January 2023

INSTRUMENTAL METHODS OF ANALYSIS

Time: 3 hours Max. Marks: 75

PART - A

(Compulsory Question)

Answer the following: (10 X 02 = 20 Marks)

	1		Answer the following: $(10 \times 02 = 20 \text{ Marks})$	
		(a)	Define Beer and Lambert's law with equation.	2M
		(b)	Discuss briefly the principle of fluorimetry.	2M
		(c)	Name different types of sampling techniques used in IR spectroscopy.	2M
		(d)	Discuss the principle of Nephelo turbidometry.	2M
		(e)	What is electrophoresis?	2M
		(f)	What are adsorption and partition column chromatography?	2M
		(g)	Discuss the applications of HPLC.	2M
		(h)	Define chromophore and auxochrome.	2M
ž.		(i)	Classify ion exchange resins.	2M
		(j)	Write applications of Gel Chromatography.	2M
			PART – B	
			(Answer any two questions: 02 X 10 = 20 Marks)	
	2		Explain with a neat schematic diagram of the instrumentation and working of a double beam UV-Vis spectrophotometer.	10M
	3		Compare and differentiate thin-layer chromatography vs radial paper chromatography in detail.	10M
	4	(a)	Summarize in detail about HPLC.	5M
		(b)	Justify the statement "HPLC is better than HPTLC".	2M
		(c)	Draw a neat and labeled diagram of HPLC.	3M
			PART – C	
(Answer any seven questions: 07 X 05 = 35 Marks)				
	5		Discuss the solvent effect in UV-Vis spectroscopy.	5M
	6		Describe photomultiplier tube with its principle and neat diagram.	5M
	7		Explain the principle and applications of flame photometry with neat labelled diagram.	5M
	8	(a)	What is the different mode of chromatograms in IR spectrophotometry?	2M
		(b)	Discuss with schematic diagram.	ЗМ
	9		Write different methods for preparations of TLC plates.	5M
	10		Discuss the different development techniques used in paper chromatography.	5M
	11		Write down the advantages and disadvantages of gas chromatography.	5M
	12		Explain the manufacture of cation exchange resin and anion exchange resin.	5M
	13		Explain the principle and theory of gel chromatography.	5M

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